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## کارگاه های آموزشی مرکز اطلاعات علمی جهاد دانشگاهی

کارگاه آنلاین  
بررسی مقابله ای متون (مقدماتی)

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کارگاه آنلاین آشنایی با پایگاه های اطلاعات علمی بین المللی و ترند های جستجو



## Effect of Smear Layer on the Push-Out Bond Strength of Two Different Compositions of White Mineral Trioxide Aggregate

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### ABSTRACT

**Introduction:** The aim of this *in vitro* study was to evaluate the effect of smear layer on the push-out bond strength of white mineral trioxide aggregate (WMTA) with and without disodium hydrogen phosphate ( $\text{Na}_2\text{HPO}_4$ ). **Materials and Methods:** Dentin discs with standard cavities were obtained from extracted human single-rooted teeth and divided to 4 groups ( $n=15$ ) according to the irrigation regimen and the canal filling material. In groups 1 and 3, canals were irrigated with normal saline; in groups 2 and 4, irrigation method included sodium hypochlorite ( $\text{NaOCl}$ ) and then ethylenediaminetetra-acetic acid (EDTA). The canals were filled with WMTA in first and second groups and with WMTA+ $\text{Na}_2\text{HPO}_4$ ; in groups 3 and 4. The samples were wrapped in wet gauze and incubated in  $37^\circ\text{C}$  for 3 days. The push-out bond strength was then measured by means of the *Universal Testing Machine* and the failure modes were examined under stereomicroscope at  $40\times$  magnification. Two-way ANOVA was used to evaluate the effect of material type and smear layer removal. Post hoc Tukey test was used for the two-by-two comparison of the groups. **Results:** The greatest and lowest mean $\pm$ standard deviation for push-out bond strength were observed in groups 4 ( $4.54\pm 1.14$  MPa) and 1 ( $1.44\pm 0.96$  MPa), respectively. The effect of removing the smear layer on the push-out bond strength of WMTA+ $\text{Na}_2\text{HPO}_4$  was significant ( $P=0.01$ ), but not for WMTA ( $P=0.52$ ). Interestingly, there was significant difference between groups 1, 3 and 2, 4 ( $P<0.05$ ). The failure mode for all experimental groups was of mixed type. **Conclusion:** Under circumstances of this *in vitro* study, removal of smear layer increases push-out bond strength when  $\text{Na}_2\text{HPO}_4$  is added to WMTA.

**Keywords:** Disodium Hydrogen Phosphate; Mineral Trioxide Aggregate; Push-Out Bond Strength; Smear Layer; Root Canal Preparation

### Introduction

Mineral trioxide aggregate (MTA) is a mixture of dicalcium silicate, tricalcium silicate, tricalcium aluminate, tetracalcium aluminoferrite, and bismuth oxide [1]. It has been widely used for perforation repair, root canal treatment of immature teeth and as a root-end filling material during apical surgery [2, 3]. MTA has met most of the requirements of an ideal root-end filling material. However, subcutaneous implantation of MTA in rats provokes severe initial reactions with coagulation necrosis and dystrophic calcification. Apart from its initial inflammatory cell reactions, its working and setting time are not ideal. [4-9]. One approach to reduce the setting time is to use disodium hydrogen phosphate ( $\text{Na}_2\text{HPO}_4$ ) as an accelerator [10]. Huang *et al.* [11], showed that 15%  $\text{Na}_2\text{HPO}_4$  solution as a liquid phase can reduce the setting time of white MTA (WMTA) to 26 min, and a diametral tensile strength of 4.9 MPa at the

initial 6-hour period can be achieved. They suggested that  $\text{Na}_2\text{HPO}_4$  solution might be an effective setting accelerator for WMTA. Ding *et al.* [12], suggested that  $\text{Na}_2\text{HPO}_4$  solution as an MTA accelerator reduces the setting time and maintains the pH value. In addition, Lotfi *et al.* showed that adding  $\text{Na}_2\text{HPO}_4$  to WMTA creates a more biocompatible material than WMTA alone [13].

Considering the clinical applications of MTA, the bond strength of material with dentin is an important factor in achieving the best sealing ability. In other words, in cases of root-end filling, the material need to remain in desired place and tolerate dislodging forces such as functional loads. The push-out bond test aims to assess the bond strength of the materials to dentin [14, 15].

The smear layer which contains organic and inorganic components is formed on dentinal walls during root canal preparation. The presence of such a layer compromises the penetration of root canal irrigants and the obturating materials into dentinal tubules, which increases the risk of bacterial

**Table 1.** The mean (SD) deviation of push-out in experimental groups

Groups	Materials	Smear layer	Strength
1	WMTA	Removed	1.44 (0.96) MPa
2	WMTA	-	2.17 (0.72) MPa
3	WMTA+Na <sub>2</sub> HPO <sub>4</sub>	Removed	2.98 (1.56) MPa
4	WMTA+Na <sub>2</sub> HPO <sub>4</sub>	-	4.54 (1.51) MPa

infection and microleakage [16]. Thus, the removal of smear layer can improve the adaptation of root canal filling materials [17]. Yildirim *et al.* showed that the apical microleakage of MTA is less in the absence of the smear layer [18].

Considering the absence of studies regarding the push-out bond strength of WMTA with and without Na<sub>2</sub>HPO<sub>4</sub>, the purpose of this *in vitro* study was to evaluate the effect of smear layer on the push-out bond strength of these two compositions of WMTA.

### Material and Methods

Sixty extracted single-rooted human maxillary anterior teeth were selected for this study. The teeth were stored in 0.5% chloramine-T. 3 mm thick slices of dentin were prepared from the mid root area using water-cooled diamond disc (SP1600 microtome; Leica, Nußloch, Germany). The lumen of the dentin slices were drilled with #2 to #5 (ISO sizes #70 to 130) Gates-Glidden burs (Dentsply, Maillefer, Ballaigues, Switzerland) to obtain 1.3 mm diameter cavities.

The samples were randomly divided into 4 groups; then they were mounted on slaps by using sticky wax to simulate the close-end model. All specimens were checked to ensure the absence of any penetration pathway for irrigation solution out of the lumen. In groups 1 and 3 irrigation was performed using 2 mL of normal saline and the cavity was dried by size #90 paper cone (Aryadent, Tehran, Iran) without contacting with canal walls. In groups 2 and 4, the standard smear layer removal method was conducted, using 2 mL 5.25% sodium hypochlorite (NaOCl) for 10 min then 2 mL normal saline followed by 17% ethylene diamine tetra acetic acid (EDTA) for 5 min were used. Two milliliters of distilled water was applied as a final flush and the canals were dried with paper cones as previously described.

In groups 1 and 2, WMTA (Tooth-colored Formula) (Dentsply, Tulsa Dental, Tulsa, OK) was mixed with distilled water according to the manufacturer's instructions and placed into the lumens. In group 3 and 4, WMTA was mixed with 2.5 wt% of Na<sub>2</sub>HPO<sub>4</sub> (Merck, Darmstad, Germany), then the powder was mixed with distilled water with a 0.3 mL/g. liquid/powder ratio. Then the lumens were filled. The samples were wrapped in gauze, wet with distilled water and incubated for 3 days in 37°C.

After the experimental time period, the samples were submitted to the push-out bond test. The push-out bond strength was measured by using *Universal Testing Machine* (Hounsfield Test Equipment, Model H5k-s, Surrey, England). The WMTA was loaded with a 1 mm diameter cylindrical stainless steel plunger at a speed of 1 mm per min. The maximum load applied to the material, which was displayed in Newton, was recorded just before dislodgment. To express

the value in MPa, the recorded amount in Newton was divided by area in mm<sup>2</sup> according to the following formula:  $2\pi r \times h$ , where the  $\pi$  is the constant 3.14,  $r$  is the root canal radius, and then  $h$  would be the thickness of the root slice in millimeters. The slices were then examined under the stereomicroscope at 40× magnification to determine the mode of bond failure. Each sample was placed into 1 of the 3 failure modes: *adhesive failure* that occurred at the material and dentin interface, *cohesive failure* that happened within the material, and *mixed failure* mode.

Two-way ANOVA was used to evaluate significance of the effect of material type and smear layer removal. Post hoc Tukey tests were used for the two-by-two comparison of the groups. Statistical significance was defined at  $P < 0.05$ . SPSS 18 statistical software was used for the analysis of data.

### Results

The greatest and lowest mean ± standard deviations for push-out bond strength were observed for groups 4 (4.54 ± 1.14 MPa) and 1 (1.44 ± 0.96 MPa), respectively (Table 1). The effect of smear layer removal on push-out bond strength of WMTA+Na<sub>2</sub>HPO<sub>4</sub> was significant (between groups 3 and 4;  $P = 0.01$ ). However, it was not significant for WMTA (between groups 1 and 2;  $P = 0.52$ ). Interestingly, there was significant difference between groups 1, 3 and 2, 4 ( $P < 0.05$ ). The failure mode for all tested groups was of mixed type.

### Discussion

This study was designed to compare the push-out bond strength of WMTA and WMTA+Na<sub>2</sub>HPO<sub>4</sub>, and also to assess the effect of smear layer removal on the dislodgment resistance of these two compositions. Endodontic materials like MTA should be able to remain in place under dislocating forces such as functional stresses caused by tooth movement during mastication or operative procedures [14, 15]. Therefore, the push-out strength of the materials that are used for vital pulp therapy, apical barrier and as root-end filling material is important to achieve successful treatment. There are different methods for measuring the adhesion of a dental material to dentin including tensile, shear, and push-out bond strength tests. The push-out test has been used in this study as a reliable method [14].

Recently some efforts were done to improve the properties of MTA. One of which was addition of Na<sub>2</sub>HPO<sub>4</sub> to reduce its setting time [11-13]. Adding Na<sub>2</sub>HPO<sub>4</sub> to WMTA can promote HA formation of hydroxyapatite, due to presence of amorphous calcium phosphate phase [11-13].

Sarkar *et al.* [19], suggested that MTA releases calcium hydroxide after mixing with water which interacts with a phosphate-containing fluid to produce calcium-deficient apatite via an amorphous calcium phosphate phase [20]. Thus, hydroxyapatite crystals nucleate and grow, filling the microscopic spaces between MTA and dentinal walls. Besides verifying the presence of HA interfacial layer, a study reported the formation of *Tag-Like Structures* (TS) extending from the intermediate layer to the dentinal tubules, similar to those reported at the resin-dentin interface. This

interlocking, improves the mechanical retention of the material that is used as an end plug in root canal space [21, 22]. Our study showed that the effect of smear layer removal was not significant even in presence of  $\text{Na}_2\text{HPO}_4$  added to WMTA. This result may be explained by the fact that WMTA cannot produce HA in presence of water. Thus, open dentinal tubules were not effectively filled by HA.

Increased push-out bond strength in the recent study after adding  $\text{Na}_2\text{HPO}_4$  to WMTA could be explained by formation of TS within dentinal tubules. However, push-out bond strength did not increase when smear layer was not removed. In other words, in spite of HA formation, smear layer acts as a barrier and prevents TS formation within dentinal tubules. In addition, Reyes-Carmona *et al.* [21-23], revealed that the formed apatite crystals deposited within collagen fibrils, which promoted the controlled mineral nucleation on dentin and triggered the formation of an interfacial layer with TS at the cement-dentin interface that improved sealing ability.

From a clinical point of view, the biomineralization process which is supported by the interaction of biomaterials with dentin in presence of phosphate groups, and removal of smear layer to open dentinal tubules; both together boost the resistance to displacement of the biomaterials from dentin. This process could be responsible for the superior dislodgement resistance in WMTA+ $\text{Na}_2\text{HPO}_4$ .

## Conclusion

Removal of smear layer increases push-out bond strength when  $\text{Na}_2\text{HPO}_4$  is added to WMTA.

Conflict of Interest: 'None declared'.

## References

- [1] Torabinejad M, Hong CU, McDonald F, Pitt Ford TR. Physical and chemical properties of a new root-end filling material. *J Endod.* 1995;21(7):349-53.
- [2] Torabinejad M, Chivian N. Clinical applications of mineral trioxide aggregate. *J Endod.* 1999;25(3):197-205.
- [3] Felipe WT, Felipe MC, Rocha MJ. The effect of mineral trioxide aggregate on the apexification and periapical healing of teeth with incomplete root formation. *Int Endod J.* 2006;39(1):2-9.
- [4] Parirokh M, Torabinejad M. Mineral trioxide aggregate: a comprehensive literature review--Part I: chemical, physical, and antibacterial properties. *J Endod.* 2010;36(1):16-27.
- [5] Yaltirik M, Ozbas H, Bilgic B, Issever H. Reactions of connective tissue to mineral trioxide aggregate and amalgam. *J Endod.* 2004;30(2):95-9.
- [6] Moretton TR, Brown CE, Legan JJ, Kafrawy AH. Tissue reactions after subcutaneous and intraosseous implantation of mineral trioxide aggregate and ethoxybenzoic acid cement. *J Biomed Mater Res.* 2000;52(3):528-33.
- [7] Vosoughhosseini S, Lotfi M, Shahi S, Baloo H, Mesgariabasi M, Saghiri MA, Zand V, Rahimi S, Ranjkesh B. Influence of white versus gray mineral trioxide aggregate on inflammatory cells. *J Endod.* 2008;34(6):715-7.
- [8] Shahi S, Rahimi S, Lotfi M, Yavari H, Gaderian A. A comparative study of the biocompatibility of three root-end filling materials in rat connective tissue. *J Endod.* 2006;32(8):776-80.
- [9] Chong BS, Pitt Ford TR. Root-end filling materials: rationale and tissue response. *Endodontic Topics.* 2005;11(1):114-30.
- [10] Kogan P, He J, Glickman GN, Watanabe I. The effects of various additives on setting properties of MTA. *J Endod.* 2006;32(6):569-72.
- [11] Huang TH, Shie MY, Kao CT, Ding SJ. The effect of setting accelerator on properties of mineral trioxide aggregate. *J Endod.* 2008;34(5):590-3.
- [12] Ding SJ, Kao CT, Shie MY, Hung Jr C, Huang TH. The Physical and Cytological Properties of White MTA Mixed with  $\text{Na}_2\text{HPO}_4$  as an Accelerant. *J Endod.* 2008;34(6):748-51.
- [13] Lotfi M, Vosoughhosseini S, Saghiri MA, Mesgariabasi M, Ranjkesh B. Effect of white mineral trioxide aggregate mixed with disodium hydrogen phosphate on inflammatory cells. *J Endod.* 2009;35(5):703-5.
- [14] Shokouhinejad N, Nekoofar MH, Irvani A, Kharrazifard MJ, Dummer PM. Effect of acidic environment on the push-out bond strength of mineral trioxide aggregate. *J Endod.* 2010;36(5):871-4.
- [15] Saghiri MA, Shokouhinejad N, Lotfi M, Aminsobhani M, Saghiri AM. Push-out bond strength of mineral trioxide aggregate in the presence of alkaline pH. *J Endod.* 2010;36(11):1856-9.
- [16] Farhad AR, Barekatin B, Koushki AR. The effect of three different root canal irrigant protocols for removing smear layer on the apical microleakage of AH26 sealer. *Iran Endod J.* 2009;3(3):62-7.
- [17] Shahrahan A, Haghdoost AA, Adl A, Rahimi H, Shadifar F. Effect of smear layer on sealing ability of canal obturation: a systematic review and meta-analysis. *J Endod.* 2007;33(2):96-105.
- [18] Yildirim T, Oruçoğlu H, Çobankara FK. Long-term evaluation of the influence of smear layer on the apical sealing ability of MTA. *J Endod.* 2008;34(12):1537-40.
- [19] Sarkar N, Caicedo R, Ritwik P, Moiseyeva R, Kawashima I. Physicochemical basis of the biologic properties of mineral trioxide aggregate. *J Endod.* 2005;31(2):97-100.
- [20] Bozeman TB, Lemon RR, Eleazer PD. Elemental analysis of crystal precipitate from gray and white MTA. *J Endod.* 2006;32(5):425-8.
- [21] Reyes-Carmona JF, Felipe MS, Felipe WT. Biomineralization ability and interaction of mineral trioxide aggregate and white portland cement with dentin in a phosphate-containing fluid. *J Endod.* 2009;35(5):731-6.
- [22] Reyes-Carmona JF, Felipe MS, Felipe WT. A phosphate-buffered saline intracanal dressing improves the biomineralization ability of mineral trioxide aggregate apical plugs. *J Endod.* 2010;36(10):1648-52.
- [23] Tay FR, Pashley DH, Rueggeberg FA, Loushine RJ, Weller RN. Calcium phosphate phase transformation produced by the interaction of the Portland cement component of white mineral trioxide aggregate with a phosphate-containing fluid. *J Endod.* 2007;33(11):1347-51.

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